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Patentanmeldung Nr. Patent application No. Demande de brevet nº

04000057.2

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Der Präsident des Europäischen Patentamts; Im Auftrag

For the President of the European Patent Office

Le Président de l'Office européen des brevets p.o.

R C van Dijk

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Anmeldung Nr:

Application no.:

04000057.2

Demande no:

Anmeldetag:

Date of filing: 05.01.04

Date de dépôt:

Anmelder/Applicant(s)/Demandeur(s):

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Bezeichnung der Erfindung/Title of the invention/Titre de l'invention: (Falls die Bezeichnung der Erfindung nicht angegeben ist, siehe Beschreibung. If no title is shown please refer to the description. Si aucun titre n'est indiqué se referer à la description.)

Acidic monoazo dyesstuffs

In Anspruch genommene Prioriät(en) / Priority(ies) claimed /Priorité(s) revendiquée(s)
Staat/Tag/Aktenzeichen/State/Date/File no./Pays/Date/Numéro de dépôt:

Internationale Patentklassifikation/International Patent Classification/Classification internationale des brevets:

C09B/

Am Anmeldetag benannte Vertragstaaten/Contracting states designated at date of filing/Etats contractants désignées lors du dépôt:

AT BE BG CH CY CZ DE DK EE ES FI FR GB GR HU IE IT LU MC NL PT RO SE SI SK TR LI

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ACIDIC MONOAZO DYESTUFFS

The invention relates to novel compounds, the use of such compounds and/or mixtures thereof as dyestuffs for printing recording materials, especially paper or papery substrates, textile fibre materials, plastic films and plastic transparencies by the inkjet printing process and also to the recording materials printed thereby.

Inkjet printing processes are becoming more and more important for industrial applications.

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Inkjet printing processes are known. In what follows, the principle of inkjet printing will only be discussed very briefly. Details of this technology are described for example in the Ink-Jet-Printing section of R.W. Kenyon in "Chemistry and Technology of Printing and Imaging Systems", Peter Gregory (editor), Blackie Academic & Professional, Chapmann & Hall 1996, pages 113-138, and references cited therein.

Together with more recently developed ink-jet ink receiving layers comprising microporous or porous surfaces the need for oxidation resistant dyestuff has become more important. The problem solved by the dystuffs of the present invention is to provide dystuffs resistant to oxidation and espescially to increase resistance to ozone. This problem has been solved by providing dyestuffs according to the invention.

This invention provides dyestuff of formula (I)

$$Z_3$$

OH HN

 R_2
 X_3
 X_4
 X_5
 X_5

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wherein

 R_1 is H; C_1 alkyl; substituted C_1 alkyl; phenyl or substituted phenyl,

R₂ is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; -COOH; -COOCH₃; -CF₃; -SO₃H, -CN or SO₂NHR₆,

where R₆ is H, C₁₋₄ Alkyl, phenyl or substituted phenyl

and

- X_1 is NR_3R_4 ; SR_5 ; OH;
- X_2 is NR_3R_4 ; SR_5 ; OH;
- 5 wherein
 - R₃ is H, C₁₋₄alkyl; substituted C₁₋₄alkyl; phenyl or substituted phenyl, naphthyl or substituted naphthyl
 - R₄ is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; phenyl or substituted phenyl, naphthyl or substituted naphthyl
- 10 R_5 is C_{14} alkyl; substituted C_{14} alkyl; phenyl or substituted phenyl and
 - is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; -OH; -COOH; -COOCH₃; -CF₃; -SO₃H; amino; alkylamino, -CN or SO₂NHR'₆, where R'₆ is H, C₁₋₄ alkyl, phenyl or substituted phenyl
- is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; OH; COOH; -SO₃H

 Z₃ is H, C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; OH; COOH; -SO₃H

 as free acid or in salt form, as well as mixtures thereof.

Preferred compounds of formula (I) are characterized in that

- 20 R₁ is H; C₁₋₄alkyl; substituted C₁₋₄alkyl,
 - R_2 is H; C_{1-4} alkyl; substituted C_{1-4} alkyl; C_{1-4} alkoxy; -COOH or -SO₃H and
 - X_1 is NR_3R_4 ; SR_5 ; OH;
 - X_2 is NR_3R_4 ; SR_5 ; OH;
- 25 wherein
 - R₃ is H, C₁₋₄alkyl; substituted C₁₋₄alkyl; phenyl or substituted phenyl, naphthyl or substituted naphthyl
 - R₄ is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; phenyl or substituted phenyl, naphthyl or substituted naphthyl
- 30 R₅ is C₁₋₄alkyl; substituted C₁₋₄alkyl; phenyl or substituted phenyl and
 - Z₁ is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; -OH; -COOH; -COOCH₃; -CF₃; -SO₃H; amino; alkylamino, -CN or SO₂NHR'₆,

where R'6 is H, C1-4 alkyl, phenyl or substituted phenyl

Z₂ is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; OH; COOH; -SO₃H

 Z_3 is H, C_{1-4} alkyl; substituted C_{1-4} alkyl; C_{1-4} alkoxy; OH; COOH; -SO₃H as free acid or in salt form, as well as mixtures thereof

Preferred compounds according to formula (I) have the formula (Ia)

$$Z_2$$
 Z_3
 $N=N$
 $N=N$
 R_2
 X_2
 X_2
 X_2
 X_3
 X_4
 X_4
 X_5
 X_5

wherein

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R_i is H; C₁₋₄alkyl; substituted C₁₋₄alkyl,

10 R_2 is H; C_{1-4} alkyl; substituted C_{1-4} alkyl; C_{1-4} alkoxy; -COOH or -SO₃H and

 X_1 is NR_3R_4 ; SR_5 ; OH;

X₂ is NR₃R₄; SR₅; OH:

wherein

15 R_3 is H, C_{1-4} alkyl; substituted C_{1-4} alkyl; phenyl or substituted phenyl, naphthyl or substituted naphthyl

R₄ is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; phenyl or substituted phenyl, naphthyl or substituted naphthyl

R₅ is C_{1.4}alkyl; substituted C_{1.4}alkyl; phenyl or substituted phenyl

20 and

Z₁ is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; -OH; -COOH; -COOCH₃; -CF₃; -SO₃H; amino; alkylamino, -CN or SO₂NHR'₆, where R'₆ is H, C₁₋₄ alkyl, phenyl or substituted phenyl

Z₂ is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; OH; COOH; -SO₃H

25 Z₃ is H, C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; OH; COOH; -SO₃H as free acid or in salt form, as well as mixtures thereof

Further more preferred compounds according to formula (I) have the formula (Ib)

$$Z_2$$
 $N=N$
 $N=N$

wherein

R₁ is H; C₁₋₄alkyl; substituted C₁₋₄alkyl,

R₂ is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; -COOH or -SO₃H

5 and

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 X_1 is NR_3R_4 ; SR_5 ; OH;

 X_2 is NR_3R_4 ; SR_5 ; OH;

wherein

R₃ is H, C₁₋₄alkyl; substituted C₁₋₄alkyl; phenyl or substituted phenyl, naphthyl or substituted naphthyl

R₄ is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; phenyl or substituted phenyl, naphthyl or substituted naphthyl

 R_5 is C_{14} alkyl; substituted C_{14} alkyl; phenyl or substituted phenyl and

is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; -OH; -COOH; -COOCH₃; -CF₃; -SO₃H; amino; alkylamino, -CN or SO₂NHR'₆, where R'₆ is H, C₁₋₄ alkyl, phenyl or substituted phenyl

Z₂ is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; OH; COOH; -SO₃H

Z₃ is H, C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; OH; COOH; -SO₃H

20 as free acid or in salt form, as well as mixtures thereof

Substituted phenyl or naphtyl means phenyl groups or naphtyl groups substituted by -SO₃H, -COOH, -OH, alkyl or alkoxy. These alkyl or alkoxy groups are by preference C₁₋₄alkyl; C₁₋₄alkoxy which may be further substituted by -SO₃H, -COOH, -OH. Preferred alkylgroups are methyl or ethyl. Preferred alkoxygroups are methoxy or ethoxy.

Substituted alkyl means alkyl groups which are further substituted by groups selected from -OH, -COOH, -NH₂, -NHalkyl, -N(alkyl)₂, -SO₃H, -O-alkyl. Alkyl groups may also be branched.

The deprotonable groups as for example the -COOH, or -SO₃H groups may also be in the salt forms e.g. -COOM, or -SO₃M. it is also possible that only a part of the protons of those deprotonable groups are neutralized. Furthermore several different cations may be present, thus the dyestuffs are in a mixed salt form.

Suitable cations M are alkali metal, alkaline earth metal, ammonium, alkanolammonium or alkylammonium cations. Examples of corresponding cations are the sodium, lithium or ammonium cations or mono-, di- or triethanolammonium cations.

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Examples of such cations are alkali metal cations, e.g. lithium, sodium, potassium, and ammonium cations or substituted ammonium cations, e.g. mono-, di-, tri- and tetramethylammonium, tri-ethylammonium, and mono-, di-, and tri-ethanolammonium. The preferred cations are the alkali metal cations and the ammonium cation, with the sodium cation being the most preferred.

A further embodiment of the present invention is the preparation of compounds of formula (I) characterized in that in a first step a compound of formula (II)

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wherein all substituents have the meanings as defined above are reacted with a compound of formula (III)

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The reaction leads to compounds according to formula (IV)

$$\begin{array}{c|c}
O & R_1 \\
O & N \\
R_2 & CI
\end{array}$$

$$HO_3S$$

$$SO_3H$$

$$(IV)$$

In a second step the product of formula (IV) is reacted with one part of a compound of formula HX_1 wherein X_1 has the formula as described above which leads to compound of formula (V)

HO₃S
$$OH$$
 HN R_2 CI V

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In a third step compound of formula (V) are condensated with a compound of formula HX_2 wherein X_2 has the formula as described above leading to compound of formula (VI)

$$\begin{array}{c|c} & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\$$

wherein substituents R_1 and R_2 have the same meanings as defined above.

In the final step a compound of formula (VI) is coupled with the diazoniumsalt of a compound of formula (VII)

$$Z_1$$
 Z_2
 Z_3
 NH_2
(VII)

15 leading to the dyestuff of formula (I)

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wherein all substituents have the same meanings as defined above.

The diazoniumsalt of the compound of formula (VII) means either a compound of formula (VIIa) or a compound of formula (VIIb)

$$Z_1$$
 Z_2
 Z_3
 Z_3
 Z_4
 Z_3
 Z_4
 Z_3
 Z_4
 Z_3
 Z_4
 Z_3
 Z_4
 Z_3
 Z_4
 Z_4
 Z_3
 Z_4
 Z_4
 Z_5
 Z_5
 Z_7
 Z_7

In the procedure of producing the compounds the starting compound of formula (II) can be synthesized by known processes, e.g. condensation of 3-nitro-benzoylchloride with H-Acid and subsequent reduction.

In the first step of the abovementioned reaction the temperature is in the range of 0°C to 40°C, preferably 0°C to 25°C and the pH is in the range of 2 to 6, preferably 3 to 5. In the second step of the abovementioned reaction the temperature is in the range of 20 - 70°C, preferably 30 to 60°C and the pH is in the range of 3 to 9, preferably 5 to 8. In the third step of the abovementioned reaction the temperature is in the range of 40 - 95°C, preferably 40 to 85°C and the pH is in the range of 5 to 10, preferably 6 to 9.

Any alkyl or alkylene group starting from C₃-alkyl or C₃-alkylene may be linear or branched. Any alkoxy group starting from C₃-alkoxy may be linear or branched.

Useful cations for salt formation include in particular alkali metal, alkaline earth metal as well ammonium cations. Examples of such cations are alkali metal cations, for example potassium, lithium or sodium ions and ammonium cations, e.g. mono-, di-, tri- or tetra-ethyl ammonium cations or mono-, di-, tri- or tetra-ethyl ammonium cations or mono-, di- or tri-ethanol ammonium cations. The cations may be the same or different, i.e. the compounds may be in mixed salt-form.

The dyestuffs according to the invention may be used alone or for shading other dyestuffs or dyestuff mixtures. The compounds (I) according to the invention may be mixed with other colorants to match exactly the desired hue, thus the colorants according to the invention. In principle any other colorant which is compatible with the compounds according to the invention may be used for this purpose.

On the other hand the compounds according to the invention having the formula (I) may itself be used as shading component for shading other colorants.

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The mixtures according to the invention itself may be used as a shading component and blended with other compatibles dyestuff mixtures to achieve the desired shade.

Instead of blending the dyestuff to the mixtures it is also possible to prepared inks as described below and mixed these inks of different shades to achieve the desired colour.

The dyestuff can also formulated into liquid compositions. Stable liquid compositions may be achieved by desalting and concentrating a dyestuff solution by passing them through separation membranes such as cellulose acetate membrane or polybenzimidazolone membrane (PBIL).

Alternatively stable solution may be produced by mixing with amines, especially with mono-, di- or tri-alkanol amines or polyethoxylated amines. Those polyethoxylated amines are normally produced by reacting ethylene oxide or propylene oxide or mixtures thereof with mono-, di- or tri-alkanol amines or with alkanol derivatives of polyamines.

The dyestuff according to the invention may alo used alone for printing. Thus in certain cases no shading dyestuffs are needed, for example if the shade is already correct.

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The amount of shading colorants are present in this mixtures is dependent on the shade that should be achieved. For example the shaded mixtures of dyes may comprise 50 - 99 wt-% of at least one compound according to formula (I) and

1 - 50 wt-% of at least one compound described as shading component as described and listed below, but not a compound according to formula (I).

By preference the shading component is present in an amount of 0.001 to 5% by weight, preferably 0.01 to 1% by weight, based on the total weight of the dry dye mixture.

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For example the shading colorant may be selected (among others) from the group consisting of C.I. Direct Red 1, 11, 37, 62, 75, 81, 87, 89, 95, 227; C.I. Acid Red 115, 131, 144, 152, 186, 245, C.I. Pigment Red 122, 176, 184, 185 and 269.

- In preferred mixtures, compounds according to formula (I) are mixed with at least one compound selected from C.I. (Colour Index) Acid Red 50, C.I. Acid Red 51, C.I. Acid Red 52, C.I. Acid Red 87, C.I. Acid Red 91, C.I. Acid Red 92, C.I. Acid Red 93, C.I. Acid Red 94, C.I. Acid Red 95, C.I. Acid Red 98 and C.I. Acid Red 289.
- In further preferred mixtures, compounds according to formula (I) are mixed with at least one compound selected from C.I. Acid Red 1, C.I. Acid Red 33, C.I. Acid Red 35, C.I. Acid Red 40, C.I. Acid Red 76, C.I. Acid Red 106, C.I. Acid Red 138, C.I. Acid Red 155, C.I. Acid Red 160, C.I. Acid Red 172, C.I. Acid Red 249, C.I. Acid Red 264 and C.I. Acid Red 265.

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In further preferred mixtures, compounds according to formula (I) are mixed with at least one compound selected from C.I. Acid Red 15, C.I. Acid Red 19, C.I. Acid Red 29, C.I. Acid Red 60, C.I. Acid Red 68, C.I. Acid Red 154 and C.I. Acid Red 176.

- In further preferred mixtures, compounds according to formula (I) are mixed with at least one compound selected from C.I. Acid Red 30, C.I. Acid Red 34, C.I. Acid Red 37, C.I. Acid Red 42, C.I. Acid Red 54, C.I. Acid Red 57, C.I. Acid Red 231, C.I. Acid Red 266, C.I. Acid Red 301 and C.I. Acid Red 337
- 30 Especially preferred mixture comprises a compound according to formula (I) and C.I. Acid Red 52 and/or C.I. Acid Red 289 which have the following formulae

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$$(H_{c}C_{2})_{3}N \longrightarrow N(C_{2}H_{5})_{2} \qquad CH_{3} \stackrel{H}{\longleftrightarrow} CH_{0}$$

$$SO_{3}N_{d} \qquad respectively.$$

It is possible to use the dyestuff (I) without any further dyestuff in the composition Ink jet printing compositions according the present invention.

A further embodiment of the present invention relates to a composition for printing recording materials, preferably paper and papery substrates, textile fibre materials, plastic films and plastic transparencies by the inkjet printing process, comprising

- 1) the dye of the formula (I) as defined above and
 - water or a medium including a mixture of water and an organic solvent, an anhydrous organic solvent or a solid having a low melting point,

The inkjet printing composition may optionally comprise further additives.

- A further embodiment of the present invention relates to a inkjet printing composition for printing recording materials, preferably paper and papery substrates, textile fibre materials, plastic films and plastic transparencies, comprising
 - 1) a mixture of dyestuffs as defined above and
- 20 2) water or a medium including a mixture of water and an organic solvent, an anhydrous organic solvent or a solid having a low melting point,

The inkjet printing composition may optionally comprise further additives.

A further embodiment of the invention relates to the use of the above mentioned compositions for the ink-jet printing process.

By additionally disposing at least one nozzle with yellow, magenta or cyan ink side by side it is possible to obtain colour reproductions in high quality. This process is known as polychromatic printing or, when three colour components are used, as trichromatic printing.

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The dyestuffs and the composition of the invention can be used with all known and suitable inkjet printers for printing paper or papery substrates, textile fibre materials, plastic films and plastic transparencies. This applies not only to the use in monochromatic printing but also to polychromatic printing, especially trichromatic printing.

The composition of the ink for the inkjet printing process has to possess a suitable conductivity, sterility in storage, viscosity and surface tension to meet the specific requirements of inkjet ink. In addition, the prints on the recording materials have to have good properties and fastness.

Useful recording materials, as mentioned above, are preferably paper and papery substrates, textile fibre materials, plastic films and plastic transparencies. But glass and metal may be used as well.

Useful papers or papery substrates include all known such materials. Preference is given to papers or papery substrates coated on at least one side with a material which is particularly receptive to ink compositions. Such papers or papery materials are described inter alia in DE 3018342, DE 4446551, EP 164196 and EP 875393.

Useful textile fibre materials are in particular hydroxyl-containing fibre materials. Preference is given to cellulosic fibre materials, which consist of or comprise cellulose. Examples are natural fibre materials such as cotton, linen or hemp and regenerated fibre materials such as, for example, viscose and also lyocell.

Useful plastic films or plastic transparencies include all known such materials. Preference is given to plastic films or plastic transparencies coated on at least one side with a material which is particularly receptive to the ink compositions. Such plastic films or plastic transparencies are described inter alia in EP 755332, US 4935307, US 4956230, US 5134198 and US 5219928.

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Further possible and useful ink jet recording elements which receive the dyestuffs or the ink compositions respectively comprise a support having thereon an image-receiving layer of micro-porous polymeric particles or porous polymeric particles or micro-porous inorganic particles or porous inorganic particles may be held together by a (polymeric) binder.

The dyes of the formula (I) used in the inks should preferably be low in salt, i.e. have a total salt content of less than 0.5% by weight, based on the weight of the dyes. Dyes having higher salt contents (owing to their preparation and/or the subsequent addition of extenders) may be desalted, for example by means of membrane separation processes, such as ultrafiltration, reverse osmosis or dialysis.

The inks preferably include a total amount of dyes which is in the range from 0.5 to 35% by weight, preferably in the range from 1 to 35% by weight, more preferably in the range from 2 to 30% by weight, most preferably in the range from 2.5 to 20% by weight, based on the total weight of the ink.

The inks include 99.5 - 65% by weight, preferably 99 - 65% by weight, more preferably 98 - 70% by weight, most preferably 97.5 - 80% by weight, of an abovementioned medium 2), which includes a mixture of water and an organic solvent, an anhydrous organic solvent or a solid having a low melting point.

When said medium 2) is a mixture including water and an organic solvent or an anhydrous organic solvent, the dye mixtures comprising at least one compounds of formula (I) are preferably completely dissolved in this medium.

Preferably the dye mixtures comprising at least one compounds of formula (I) have a solubility of not less than 2.5% by weight in this medium 2) at 20°C.

When the ink composition of the invention is used for printing paper or papery substrates, the inks are preferably used together with the following compositions.

When the medium is a mixture of water and an organic solvent, the weight ratio of water to organic solvent is preferably in the range from 99:1 to 1:99, more preferably in the range from 99:1 to 50:50, particularly preferably in the range from 95:5 to 80:20.

It is preferable for the organic solvent, which is included in the mixture with water to be 5 a water-soluble solvent or a mixture of various water-soluble solvents. Preferred watersoluble organic solvents are C₁₋₆-alcohols, preferably methanol, ethanol, n-propanol, isopropanol, n-butanol, sec-butanol, tert-butanol, n-pentanol, cyclopentanol and cyclohexanol; linear amides, preferably dimethylformamide, or dimethylacetamide; ketones and keto alcohols, preferably acetone, methyl ethyl ketone, cyclohexanone and 10 diacetone alcohol; water-miscible ethers, preferably tetrahydrofuran and dioxane; diols, preferably diols possessing 2 to 12 carbon atoms, e.g. 1,5-pentanediol, ethylene glycol, propylene glycol, butylene glycol, pentylene glycol, hexylene glycol and thiodiglycol and oligo- and poly-alkylene glycols, preferably diethylene glycol, triethylene glycol, 15 1,2-propylenglycol, polyethylene glycol and polypropylene glycol; triols, preferably glycerol and 1,2,6-hexanetriol; mono-C1-4-alkyl ethers of diols, preferably mono-C1-4alkyl ethers of diols possessing 2 to 12 carbon atoms, particularly preferably 2methoxyethanol, 2-(2-methoxyethoxy)ethanol, 2-(2-ethoxyethoxy)ethanol, 2-[2-(2-ethoxyethoxy)ethanol, 2-[2-(2-ethoxyethoxyethoxy)ethanol, 2-[2-(2-ethoxyethox methoxyethoxy]-ethanol, 2-[2-(2-ethoxyethoxy)ethoxy]ethanol, diethylenglycol-20 mono-n-butylether, ethylene glycol monoallyl ether and polyoxyethylenalkylether (for example Emulgen 66 of KAO Corp. (Emulgen is a trade mark of the Kao Corp.)); alkanolamines, preferably 2-diethylamine-1-ethanol, 3-dimethylamine-1-propanol, 3diethylamine-1-propanol, 2-(2-aminoethoxy)ethanol, 2-(2-dimethylaminoethoxy)ethanol, 2-(2-diethylaminoethoxy)ethanol, mono-, di-, triethanolamine, monoglycolamines and polyglycolamines, which may be obtained by reaction of 25 ammonia, alkyl- or hydroxyalkylamines like methylamine, ethylamine, dimethylamine, diethylamine, mono-, di- and triethanolamines with alkyleneoxides for example ethylenoxide, 1,2-propylenoxide, 1,2-butylenoxide or 2,3-butylenoxide in suitable ratios as described in DE2061760A, preferably diethylenglycolamine, triethylenglycolamin, 30 Bis-diethylenglycolamin, polyoxyethylen-(6)-triethanolamine, polyoxyethylen-(9)triethanolamine, o-(2-aminoethyl)-polyethylenglycol 750, o,o-Bis-(2-aminopropyl)polyethylenglycol 500, 800, 1900, 2000, o,o'-Bis-(3-aminopropyl)-polyethylenglycol 1500, cyclic amides, preferably 2-pyrrolidone, N-methyl-2-pyrrolidone, N-ethyl-2-

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pyrrolidone, N-(2-hydroxy)ethyl-2-pyrrolidone, caprolactam and 1,3-dimethylimidazolidone; cyclic esters, preferably caprolactone; sulphoxides, preferably dimethyl sulphoxide and sulpholane.

In a preferred composition, the medium as per 2) includes water and at least 2 or more, more preferably 2 to 8, water-soluble organic solvents.

Particularly preferred water-soluble solvents are cyclic amides, particularly 2pyrrolidone, N-methyl-2-pyrrolidone and N-ethyl-2-pyrrolidone; N-(2-hydroxy)ethyl-2pyrrolidone, C_{1.6}-alcohols, preferably n-propanol, cyclohexanol, diols, preferably 1.5pentanediol, ethylene glycol, thiodiglycol, diethylene glycol, triethylene glycol and 1.2propyleneglycol, triols, preferably glycerol; and mono-C_{1.4}-alkyl and C_{1.4}-alkyl ethers of diols, more preferably mono-C1-4-alkyl ethers of diols possessing 2 to 12 carbon atoms, particularly preferably 2-[2-(2-methoxyethoxy)-ethoxy]-ethanol. diethylenglycol-mono-n-butylether, (for example Emulgen 66 of KAO Corp. (Emulgen is a trade mark of the Kao Corp.)), 2-diethylamine-1-ethanol, 3-dimethylamine-1-3-diethylamine-1-propanol, propanol, 2-(2-diethylaminoethoxy)-ethanol, triethanolamine, diethylenglycolamin, polyglycolamines, preferably polyoyxethylen-(6)-triethanolamin, polyoxyethylen-(9)-triethanolamine, o-(2-aminoethyl)-polyethylen 750, o,o-bis-(2-aminopropyl)-poyethylen o,o-bis-(3-aminopropyl)-500 and polyethylenglycol 1500.

A preferred medium as per 2) comprises

- (a) 75 to 95 parts by weight of water and
- (b) 25 to 5 parts of one or more of the watersoluble solvents. wherein the parts are by weight and all parts of (a) and (b) add up to 100.

Examples of further useful ink compositions including water and one or more organic solvents are found in the Patent Specifications US 4963189, US 4703113, US 4626284 and EP 425150A.

When the medium as per 2) includes an anhydrous (i.e. less than 1% by weight of water) organic solvent, this solvent will have a boiling point of 30 to 200°C, more preferably of 40 - 150°C, particularly preferably of 50 - 125°C.

5 The organic solvent can be water-insoluble, water-soluble or mixtures of such solvents.

Preferred water-soluble organic solvents are all above-described water-soluble organic solvents and mixtures thereof.

Preferred water-insoluble solvents include inter alia aliphatic hydrocarbons; esters, preferably ethyl acetate; chlorinated hydrocarbons, preferably CH₂Cl₂; and ethers, preferably diethyl ether; and mixtures thereof.

When the liquid medium as per 2) includes a water-insoluble organic solvent, it is preferable to add a polar solvent to increase the solubility of the dye in the liquid medium.

Examples of such polar solvents are C_{1-4} -alcohols, preferably ethanol or propanol; ketones, preferably methyl ethyl ketone.

The anhydrous organic solvent can consist of a single solvent or a mixture of 2 or more different solvents.

When it is a mixture of different solvents, a mixture including 2 to 5 different anhydrous solvents is preferred. This makes it possible to provide a medium as per 2) which permits good control of the drying properties and of the stability of the ink composition in storage.

Ink compositions including an anhydrous organic solvent or mixtures thereof are of particular interest when rapid drying times are required and especially when they are used for prints on hydrophobic and non-absorbing substrates, such as plastic, metal and glass.

Preferred low-melting media have a melting point of 60 to 125°C. Useful low-melting solids include long-chain fatty acids or alcohols, preferably those having a C₁₈₋₂₄-carbon chain, and sulphonamides.

The ink composition of the invention may further include as auxiliaries additional components which are normally used in inkjet inks, for example viscosity improvers, surface tension improvers, biocides, corrosion inhibitors, levelling agents, drying agents, humefactants, ink penetration additives, light stabilizers, UV absorbers, optical brighteners, coagulation reducers, ionic or nonionic surfactants, conducting salts and pH buffers.

These auxiliaries are preferably added in an amount of 0 - 5% by weight.

To prevent precipitation in the ink compositions of the invention, the dyes used have to be purified clean. This can be done with commonly known purifying methods.

When printing textile fibre materials, useful additives, as well as the solvents, include water-soluble nonionic cellulose ethers or alginates.

- The preferred material is paper. The paper may be plain or treated.

 Preference is given to ink compositions having a viscosity of 1 to 40 mPa's, especially 5 to 40 mPa's, preferably 10 to 40 mPa's. Ink compositions having a viscosity of 10 to 35 mPa.s are particularly preferred.
- 25 Preference is given to ink compositions having a surface tension of 15 73 mN/m, especially 20 65 mN/m, particularly preferably 30 50 mN/m.

Preference is given to ink compositions having a conductivity of 0.1 - 100 mS/cm, especially 0.5 - 70 mS/cm, particularly preferably 1.0 - 60 mS/cm.

The inks may further include buffer substances, for example borax, borate or citrate. Examples are sodium borate, sodium tetraborate and sodium citrate.

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They are used in particular in amounts of 0.1 to 3% by weight, preferably 0.1 to 1% by weight, based on the total weight of the ink, to set a pH of for example 5 to 9, especially 6 to 8. A citrate buffer is preferred in the case of alginatic inks.

The inks may further include customary additives, for example foam suppressants or especially fungal and/or bacterial growth inhibitors. These are customarily used in amounts of 0.01 to 1% by weight, based on the total weight of the ink.

The printing inks and also the dye mixtures comprise at least the compound of formula 10 (I).

The prints obtainable by the process of the invention have good general fastnesses, very good ozone-fastness, very good lightfastness and also sharp contours and a high colour strength. The inks provide prints of high optical density. Especially good brilliance together with a good ozone and light fastness is achieved with the dyes according to the invention.

The printing inks used are notable for good stability and good viscosity properties. The recording fluids of the invention have viscosity and surface tension values which are within the ranges suitable for the ink-jet processes. The viscosity remains virtually unchanged even in the event of high shearing forces occurring during printing.

Recording fluids according to the invention in storage are not prone to the formation of precipitates that leads to fuzzy prints or nozzle cloggage.

A further aspect of the present invention is the use of the printing ink in trichromatic printing. Trichromatic printing is a very large application for all recording materials. This form of printing is normally carried out with a yellow, red and blue ink composition. Furthermore, the magenta dye mixtures of the invention may be used as an ink set in combination with black, yellow and/or cyan recording fluids.

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This invention further provides recording materials, which have been printed with a composition according to the invention.

The compounds according to the invention may be used for dyeing cationic dyeable materials such as: homo- or mixed-polymers of acrylonitrile, acid modified polyester or polyamide; wool; leather including low affinity vegetable-tanned leather; cotton; bast fibers such as hemp, flax, sisal, jute, coir and straw; regenerated cellulose fibers, glass or glass products comprising glass fibers; and substrates comprising cellulose for example paper and cotton. They may also be used for printing fibers, filaments and textiles comprising any of the above mentioned materials in accordance with known methods. Printing may be effected by impregnation of the material to be printed with a suitable printing paste comprising one or more compounds of the present invention. The type of printing paste employed, may vary depending on the material to be printed. Choice of a suitable commercially available printing paste or production of a suitable paste, is routine for one skilled in the art. Alternatively, as already described, the compounds of the present invention may be used in the preparation of inks suitable for example for jet printing, in accordance with conventional methods as explained above.

Moreover, the dyestuffs are used for dyeing or printing of paper e.g., sized or unsized, wood-free or wood-containing paper or paper-based products such as cardboard. They may be used in continuous dyeing in the stock, dyeing in the size press, in a conventional dipping or surface coloring process. The dyeing and printing of paper is effected by known methods.

The dyeings and prints and particularly those obtained on paper, show good fastness properties.

The paper dyeings or printings made with the compounds according to the invention are clear and brilliant and have good light fastness. On exposure to light for a long time, the shade of the dyeing fades tone in tone. They show very good wet fastness properties; being fast to water, milk, fruit juice, sweetened mineral water, tonic water, soap and sodium chloride solution, urine etc. Furthermore, they have good alcohol fastness properties. The wet fastness properties are improved compared to known dyes showing

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otherwise similar properties. They do not exhibit a tendency towards two-sidedness. It is important to mention the extreme good fastness against ozon.

Paper dyed or printed with the compounds of the present invention can be bleached either oxidatively or reductively, a feature, which is important for the recycling of waste paper and old paper products.

The compounds of the present invention may also be used to dye paper containing wood-pulp where even dyeings, having good fastness properties are obtained. Furthermore, they may be used for the production of coated paper in accordance with known methods. Preferably when coating, a suitable filler, for example kaolin, is employed in order to give a one-side coated paper.

Moreover, the dye mixtures of the invention are useful as colorants in the electrophotographic toners and developers, for example one- and two-component powder toners, magnetic toners, liquid toners, polymerization toners and other specialty toners.

Typical toner binders are addition polymerization, polyaddition and polycondensation resins, such as styrene, styrene-acrylate, styrene-butadiene, acrylate, polyester, phenolic and epoxy resins, polysulfones, polyurethanes, individually or in combination, and also polyethylene and polypropylene, in or to which further ingredients, such as charge control agents, waxes or flow agents, may be present or added subsequently. Dye mixtures according to the invention are further useful as colorants in powders and powder coating materials, especially triboelectrically or electrostatically sprayed powder coating materials, which are used to coat the surfaces of articles made for example of metal, wood, plastic, glass, ceramic, concrete, textile material, paper or rubber. Powder coating resins employed are typically epoxy resins, carboxyl- and hydroxyl-containing polyester resins, polyurethane resins and acrylic resins together with customary curing agents. Combinations of resins are also used. For instance, epoxy resins are frequently used in combination with carboxyl- and hydroxyl-containing polyester resins.

The dye mixtures of the invention are also useful as colorants for color filters, for additive as well as subtractive color generation (P. Gregory "Topics in Applied Chemistry: High Technology Applications of Organic Colorants" Plenum Press, New York 1991, page 15-25), and also as colorants in electronic inks for electronic newspapers.

The examples hereinbelow illustrate the invention. Temperatures are in degrees Celsius; parts and percentages are by weight, unless otherwise stated.

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Examples of ink compositions:

A preferred ink composition according to the invention comprises

0.5 - 35 parts of a compound of formula (I),

65 - 99.5 parts of water or a medium including a mixture of water and an organic solvent, an anhydrous organic solvent or a solid having a low melting point and optionally

0 - 5 parts of one or more additives.

20 A more preferred ink composition according to the invention comprises

1 - 20 parts of a compound of formula (I),

80-99 parts of water or a medium including a mixture of water and an organic solvent, an anhydrous organic solvent or a solid having a low

melting point and optionally

25 0 - 5 parts of one or more additives.

A particularly preferred ink composition according to the invention comprises

1 - 5 parts of a compound of formula (I),

95-99 parts of water or a medium including a mixture of water and an organic

solvent, an anhydrous organic solvent or a solid having a low

melting point and optionally

0 - 5 parts of one or more additives.

	A more preferred in	nk composition according to the invention comprises
	0.5 - 35 parts	of a dye mixtures comprising at least one compound of formula (I)
		and at least C.I. Acid Red 52 or C.I. Acid Red 289, and
5	65 - 99.5 parts	of water or a medium including a mixture of water and an organic
		solvent, an anhydrous organic solvent or a solid having a low
		melting point and optionally
	0 - 5 parts	of one or more additives.
10	A particularly prefe	rred ink composition according to the invention comprises
	1 - 20 parts	of a dye mixtures comprising at least one compound of formula (I)
		and at least C.I. Acid Red 52 or C.I. Acid Red 289, and
	80-99 parts	of water or a medium including a mixture of water and an organic
	•	solvent, an anhydrous organic solvent or a solid having a low
15		melting point and optionally
	0 - 5 parts	of one or more additives.
	A further preferred i	nk composition according to the invention comprises
	1 - 5 parts	of a dye mixtures comprising at least one compound of formula (I)
20		and at least C.I. Acid Red 52 or C.I. Acid Red 289, and
	95 -99 parts	of water or a medium including a mixture of water and an organic
		solvent, an anhydrous organic solvent or a solid having a low
		melting point and optionally
	0 - 5 parts	of one or more additives.
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The sum total of all the parts of the abovementioned compositions according to the invention is 100 parts.

The above mentioned composition is preferably prepared by heating the medium to 30 - 40°C and then adding a dye of a compound of formula (I) or a dye mixtures comprising at least one compound of formula (I) and at least C.I. Acid Red 52 or C.I. Acid Red 289. The composition is then cooled down to room temperature.

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This ink composition is preferably used for printing papers or papery substrates.

The following Examples further serve to illustrate the invention, without restricting the scope of protection to these Examples. In the Examples all parts and all percentages are by weight, and the temperatures given are in degrees Celsius, unless indicated to the contrary.

Example 1

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a) Condensation of 1-(3'-aminobenzoylamino)-8-hydroxynaphthalin-3,6-disulfonic acid with Cyanurchloride

220 g of 1-(3'-aminobenzoylamino)-8-hydroxynaphthalin-3,6-disulfonic acid are solved with sodiumhydroxide in water. This solution were added to a suspension of 92,2 g of cyanurchloride in ice/water mixture, keeping the pH at 3 by adding sodiumhydroxide.

After the termination of this reaction the compound of formula (VIII)

is obtained.

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b) Condensation of compound VIII with anthranilic acid

68,6 g of anthranilic acid is solved in water with sodiumhydroxide and added to the solution of compound (VIII). The temperature was increased to 45-47°C and the pH keeped at 7.

After the termination of the reaction compound of formula (IX) is obtained. The compound was isolated by precipitation with sodiumchlorid.

c) Condensation of IX with 3-Mercapto-1-propanesulfonic acid

5 562,5 g of presscake of compound (IX) was solved in water and 86,5 g of 3-Mercapto-1-propanesulfonic acid were added. The temperature was increased to 60°C and the ph keeped at 7,5-8,0

After 5 hours a compound of formula (X)

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is salted out and filtered off.

d) Diazotization and coupling with 2-amino-1-carboxybenzene-5-sulfonic acid

100g of 2-amino-1-carboxybenzene-5-sulfonic acid in water are diazotizated at 0-5°C and then added to a solution of 600 g presscake of compound (X) in water, keeping the pH at 7 by adding sodium hydroxide. After termination of this coupling the dyestuff of formula (XI)

is salted out and filtred.

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The presscake is washed with brain and then solved in water. The solution is desalted by membranfiltration at temperatures of 25-50°C and pressures of 10-30 bars. After concentration and drying a dark red powder with a λ_{max} of 545,3 nm (measured in 2 wt- % sodium acetate in H₂O at room temperature) is obtained.

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Example 2

- a) Condensation of compound VIII with 2-amino-1-hydroxybenzene-4-sulfonic acid
 - 95 g of 2-amino-1-hydroxybenzene-4-sulfonic acid is solved in water with sodiumhydroxide and added to a solution of compound (VIII), as produzed in example 1. The temperature was increased to 30-35°C and the pH keeped at 5.
- 15 After the termination of this reaction the compound of formula (XII)

is obtained.

20 b) Condensation of compound (XII) with N-methyl-taurine

600 g presscake of compound XII was solved in water and 70 g of N-methyl-taurine were added. The temperature was increased to 40-50°C and the pH keeped at 8-8.5.

A compound of formula (XIII)

25 is obtained.

d) Diazotization and coupling with 4-amino-1-methylbenzene-3-sulfonic acid

88g of 4-amino-1-methylbenzene-3-sulfonic acid in water are diazotizated at 0-5°C and then added to a solution of 580 g presscake of compound (XIII) in water, keeping the pH at 6-7 by adding sodium hydroxide. After termination of this coupling the dyestuff of formula (XIV) is salted out and filtred. The subsequent membrane filtration (at temperatures of 25-50°C and pressures of 10-30 bars) and drying leads to a dark red powder. The dyestuff of formula (XIV)

has a λ_{max} of 545,7 nm (measured in 2 wt-‰ sodium acetate in H₂O at room temperature), is obtained.

Example 3

20 a) Condensation of N-methyl-taurine with compound IX

138 g of the moist presscake consisting of the compound of formula (IX) is added to 500 ml of water. 17,1 g N-methyl-taurine is added and the temperature is increased to 60-65°C and the pH-value is keeped at 8-8,5.

After 2 hours a compound of formula (XV)

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is obtained. This product is salted out and filtred off.

5 d) Diazotization and coupling of compound (XV) with 2-naphthylamine-1,6disulfonic acid

37 g of 2-naphthylamine-1,6-disulfonic acid in water are diazotizated at 0-5°C and then added to a solution of 250 g presscake of compound (XV) in water, keeping the pH at 7-8,5 by adding sodium hydroxide. After termination of this coupling the dyestuff of formula (XVI) is salted out and filtred. The subsequent membrane filtration (at temperatures of 25-50°C and pressures of 10-30 bars) and drying leads to a dark red powder. The dyestuff of formula (XIV)

has a λ_{max} of 545,5 nm (measured in 2 wt-‰ sodium acetate in H₂O at room temperature),

20 <u>TABLE 1</u> Examples 4 - 106

Analogously to Example 1 the following compounds according to formula (XVII) were obtained.

$$\begin{array}{c|c} DK & OH & HN & HN & X_1 \\ N=N & N & N & X_2 \\ HO_3S & SO_3H & X_2 \end{array}$$
 (XVII)

Exp.	DK	X ₁	X ₂	λ_{max}
4	но,ѕ соон	NH COOH	- ИН	545,2
5	но, соон	NH COOH	→ NOH	545,9
6	но,ѕ соон	NH COOH	* N_O	546,1
7	HO ₃ S COOH	NH COOH	•NOH	546,4
8	но _з ѕ соон	NH	NH SO ³ H	545,2
9	но, я	NH COOH	_^N	545,8
10	но, соон	NH COOH	NH. COOH	546,8
11	но ₃ ѕ соон	• NH COOH	s соон	545,8
12	но,з соон	· NH COOH	• он sон	546,1
13	но,ѕ	• NH. COOH	НООН	545,4
14	но _з е соон	NH SO ₃ H	*SSO ₃ H	545,5

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Exp.	DK	X ₁	. X ₂	λ_{max}
15	но ₃ ѕ Соон	NH OH	- NНОН	545,4
16	но _з ѕ соон	OH -NH SO ₃ H	он Он	545,7
17	но, соон	OH SO ₃ H	* N_O	546,2
18	но₃ѕ соон	*NH SO3H	, N	546,2
19	но₃ѕ соон	*NH SO3H	NH COOH	546,3
20	но, соон	NH OH	sсоон	546,3
21	HO ₃ S COOH	HO ₃ S COOH	*SSO ₃ H	545,7
22	но,,ѕ соон	HO ₃ S COOH	N SO ₃ H	545,8
23	но _з ѕ соон	HO ₃ S COOH	* N O	546,8
24	но,5 соон	HO ₃ S COOH	NH. O OH	544,9
25	но _з ѕ соон	• S	• NH. ✓ соон	543,6
26	но,5 соон	• S COOH	N SO ₃ H	543,0

Exp.	DK	X ₁	X ₂	λ_{\max}
27	но, соон	· 8	• N_O	542,9
28	но, соон	· s	NHООН	543,2
29	но, соон	*SSO ₃ H	°sso₃H	545,7
30	HO ₃ S COOH	*sso₃H	N SO ₃ H	545,5
31	но₃ѕ Соон	-NH COOH	* \$\$O ₃ H	546,0
32	но₃ѕ	NH	sсоон	545,8
33	но,	NH COOH	• NH OH	546,7
34	HO ₃ S COOH	NH COOH	+ N_O	545,5
35	но _з ѕ соон	* NH COOH	_Nso₃H	546,3
36	но,ѕ соон	NH COOH	NH. COOH	546,5
37	но, в	NH	NHOOH	545,8
38	но₃ѕ .	* NH SO ₃ H	*SSO ₃ H	546,6
39	но _з ѕ .	NH OH SO ₂ H	\$ СООН	546,4

Exp.	DK	X ₁	X ₂	λ_{max}
40	но₃ѕ Соон	OH SO ₃ H	*N_O	546,5
41	но _з ѕ соон	*NH OH	, N SO ³ H	545,9
42	но _з ѕ соон	· OH * NH SO _a H	№соон	546,4
43	но _з ѕ соон	- S COOH	NH. ∕О ∕ОН	545,4
44	но з соон	· s	NSO3H	545,0
45	но,35	- s	NH COOH	545,7
46	но, в	· s	* N_O	545,2
47	но в	*SSO ₃ H	`sso ₃ H	546,3
48	но,	* SSO ₃ H	✓N~✓SO ₉ H	545,0
49	но, в	*sso,н	NH-VCOOH	545.6
50	но, в	*SSO ₂ H	• N O	545,6
51	HN S COOH	* NH	*SSO ₃ H	530,0
52	HN S COOH	* NH COOH	*N_0	513,4

Exp.	DK	X ₁	X ₂	λ_{\max}
53	ни в соон	*NH COOH	N_SO ₃ H	518,4
54	ни соон	· NH COOH	s OH	515,1
55	HN S COOH	NH COOH	NH COOH	518,3
56	HN S COOH	NH OH SO ₃ H	°5SO ₃ H	520,3
57	HN S COOH	· NH SO ₃ H	OH S OH	519,5
58	HN S COOH	· NH SO3H	+ N_O	518,8
59	ни соон	· NH SO3H	N_√SO₃H	519,3
60	HN COOH	HO ₃ S COOH	SO _s H	520,2
61	ни соон	HO ₃ S COOH	NSO ₃ H	520,3
62	HN S COOH	°SSO ₃ Н	*SSO ₃ H	520,3
63	HN S COOH	• NH	*sso ₃ H	526,8

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Exp.	DK	X ₁	X ₂	λ_{\max}
64	нх соон	· NH COOH	N SO ₃ H	527,3
65	HN S COOH	COOH - NH	* N_O	528,1
66	HN S COOH	*NH SO ₃ H	*sso ₃ H.	527,6
67	HN COOH	NH OH SO ₃ H	* sсоон	528,0
-68	HN COOH	°S,SO₃H	*sso,H	517,8
69	СООН	NH •	* NO	545,6
70	Соон	NH *	- NH_OH	545,8
71 -	Соон	· COOH	,N	546,0
72	соон	COOH	s OH	546,1
73	Соон	COOH NH .	scooh	545,8
74	СООН	COOH NH .	*SSO ₃ H	545,7
75	ССООН	HO ₃ S COOH	_N.	545,9

Exp.	DK	X ₁	X_2	λ_{\max}
76	Соон	OH "NH SO ₂ H	**	546,3
77	ССООН	OH SO ₃ H	ин о он	546,6
78	Соон	SO ² H	s OH S	546,0
79	Соон	•NH •SO ₃ H	*sso₃H	545,8
80	SO ₃ H	NH *	,N	537,8
81	SO ₃ H	NH -	* O	538,0
82	SO ₃ H	NH +	s OH	538,2
83	SO ₃ H	NH +	NH. COOH	537,9
84	SO ₃ H	COOH NH *	″ѕѕо₃н	536,4
85	SO ₃ H	•NH •SO ₃ H	* N_O	537,6
86	SO ₃ H	NH SO ₃ H	^N~~so₃H	536,4
87	SO ₃ H	NH SO ₃ H	, NHСООН	536,5

Exp.	DK	X_1	X ₂	λ_{max}
88	SO₃H.	• NH SO ₃ H	* s-⁄_соон	538,0
89	SO ₃ H	NH SO ₃ H	*sso ₃ н	538,3
90	SO₃H	COOH	N SO ₃ H	546,0
91	SO ₃ H	NH *	• N_O	546,7
92	SO ₃ H	NH +	*SSO ₃ H	546,3
93	SO ₃ H	NH SO ₃ H	• N_O	546,7
94	SO ₃ H	NH SO ₃ H	, NH СООН	546,5
95	SO ₃ H	*sso,H	*SSO ₃ H	545,9
96	O SO ₃ H	NH.	°SSO₃H	558,9
97	SO ₃ H	NH.	, NHСООН	559,0
98	SO ₃ H	NH.	*SSO ₃ H	556,2
99	SO ₃ H	NH.	s OH	562,1
100	SO ₉ H	NH.	• N O	560,1
101	SO ₃ H	• NH SO ₃ H	NH СООН	559,0

Exp.	DK	. X ₁	X ₂	λ_{\max}
102	SO ₃ H	"NH SO ₃ H	, м	559,4
103	СООН	COOH NH.	*SSO ₃ H	547,3
104	соон	COOH NH .	NSo₃H	547,4
105	соон	NH SO ₃ H	*sso ₃ H	547,7
106	соон	SO3H	,NSO ₃ H	547,0

 $^{^*\}lambda_{\text{max}}$ -values are measured in 2 wt-‰ sodium acetate in H₂O at room temperature)

TABLE 2 Examples 107-118

Analogously to Example 3 the following compounds according to formula (XVIII) were obtained.

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Exp.	DK	X ₁	X ₂	λ _{max} * [nm]
107	50 ₃ Н	ин.	°SSO₃H	550,8

Exp.	DK	X ₁	X ₂	λ _{max} " [nm]
108	SO₃H	COOH NH.	NH COOH	549,8
109	So₃H	NH •	s OH	551,1
110	so ₃ н	NH.	*NO	550,5
111	SO ₃ H	• NH SO ₉ H	· . √N ∕ SO₃H	553,2
112	SO ₃ H	*NH SO ₃ H	NH. COOH	550,8
113	SO ₃ H	COOH NH .	*sso ₃ H	546,6
114	SO₃H SO₃H	COOH NH -	*N_O	545,8
115	SO,H SO,H	COOH NM .	NH COOH	545,1
116	\$0 ₃ H	СООН	oH s_OH	546,8
117	SO,H SO,H	· NH OH SO ₃ H	`N~~`\$0³H	545,5
118	SO ₃ H	NH SO ₃ H	NHсоон	544,3

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 λ_{max} -values are measured in 2 wt-% sodium acetate in H₂O at room temperature)

Dyestuff mixtures

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The following dyestuff mixtures are used to formulate a composition for printing recording materials as described above. Such a composition comprises

- 1) a dyestuff or a mixture of dyestuffs as defined above and
- 2) water or a medium including a mixture of water and an organic solvent, an anhydrous organic solvent or a solid having a low melting point, and
 - 3) optionally further additives.

15 Example 119 (Dyestuff mixture 1)

90 wt-% of the Example 1 and 10 wt-% of C.I. Acid Red 52.

20 This mixture is used to formulate a composition comprising

2.5 wt-% of Example 119 (dyestuff Mixture 1)

15 wt-% of N-methyl-pyrrolidone

82.5 wt-% of water.

25 This mixture is further used to formulate a composition comprising

2.5 wt-% of Example 119 (dyestuff Mixture 1)

20 wt-% of 1,2 propyleneglycol

77.5 wt-% of water:

30 TABLE 3 / Examples 120 - 123

Compound 1 is a compound according to formula (I) or a mixture thereof. Compound 2 is a dyestuff named by its Colour Index name.

Exp.	Compound 1	wt-% of	Compound 2	wt-% of	
		Compound 1	·	Compound 2	
120	Example 1	80	C.I. Acid Red 52	20	
121	Example 1	70	C.I. Acid Red 289	30	
122	Example 2	80	C.I. Acid Red 52	20	
123	Example 2	70	C.I. Acid Red 289	30	

All Examples (1-118) as well as the compounds according to formula (I) and mixtures thereof are suitable as Compound 1.

All compounds C.I. Acid Red and C.I. Direct Red compounds as well as mixtures thereof, which are listed in the description are suitable as Compound 2

APPLICATION EXAMPLES

APPLICATION EXAMPLE A

An ink consisting of 2.5 parts of the dyestuff of Example 1 in 97.5 parts of a mixture of water and 2-pyrrolidone where the ratio of water to N-methylpyrrolidone is 85:15 is introduced into an HP 880C DeskJet Printer and printed onto an A4 Epson Premium Glossy Photo Paper; S041287 (HP and DeskJet are registered trademarks of Hewlett-Packard, Palo Alto, California, USA; Epson is a registered trademark of Seiko Epson Kabushiki Kaisha.). The magenta prints thus obtained have very good ozon- and lightfastnesses.

This application example can be used in a similar manner for all the examples of the present application. Similarly, mixtures of individual dyes can be used. The prints thus obtained have very good ozon- and lightfastnesses.

APPLICATION EXAMPLE B

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An ink consisting of 2.5 parts of the dyestuff of Example 2 in 97.5 parts of a mixture of water, propylene glycol and isopropanole where the ratio of water: propylene glycol and isopropylene is 90:5:5 is introduced into an HP 880C DeskJet Printer and printed onto A4 Epson Premium Glossy Photo Paper; S041287 (HP and DeskJet are registered trademarks of Hewlett-Packard, Palo Alto, California, USA; Epson is a registered trademark of Seiko Epson Kabushiki Kaisha.). The magenta prints thus obtained have very good ozon- and lightfastnesses.

This application example can be used in a similar manner for all the examples of the present application. Similarly, mixtures of individual dyes can be used. The prints thus obtained have good fastnesses.

30 APPLICATION EXAMPLE C

An ink consisting of 2.5 parts of the dyestuff of Example 3 in 97.5 parts of a mixture of water and 2-pyrrolidone where the ratio of water to N-methylpyrrolidone is 85:15 is introduced into an HP 880C DeskJet Printer and printed onto an A4 Epson Premium

Glossy Photo Paper; S041287 (HP and DeskJet are registered trademarks of Hewlett-Packard, Palo Alto, California, USA; Epson is a registered trademark of Seiko Epson Kabushiki Kaisha.). The magenta prints thus obtained have good fastnesses especially ozon- and lightfastness.

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This application example can be used in a similar manner for all the examples of the present application. Similarly, mixtures of individual dyes can be used. The prints thus obtained have good fastnesses and the color is brilliant.

10 APPLICATION EXAMPLE D

An ink consisting of 2.5 parts of dyestuff of Example 14 in 97.5 parts of a mixture of water, propylene glycol and isopropanole where the ratio of water: propylene glycol and isopropylene is 90:5:5 is introduced into an HP 880C DeskJet Printer and printed onto an A4 Epson Premium Glossy Photo Paper; S041287 (HP and DeskJet are registered trademarks of Hewlett-Packard, Palo Alto, California, USA; Epson is a registered trademark of Seiko Epson Kabushiki Kaisha.). The magenta prints thus obtained have good fastnesses especially ozon- and lightfastness and the color is brilliant.

This application example can be used in a similar manner for all the examples of the present application. Similarly, mixtures of individual dyes can be used. The prints thus obtained have good fastnesses and the colors is brilliant.

APPLICATION EXAMPLE E

An ink consisting of 2.5 parts of the dyestuff of Example 42 in 97.5 parts of a mixture of water and 2-pyrrolidone where the ratio of water to N-methylpyrrolidone is 85:15 is introduced into an HP 880C DeskJet Printer and printed onto an A4 Epson Premium Glossy Photo Paper; \$041287 (HP and DeskJet are registered trademarks of Hewlett-Packard, Palo Alto, California, USA; Epson is a registered trademark of Seiko Epson Kabushiki Kaisha.). The magenta prints thus obtained have very good ozon- and lightfastnesses.

This application example can be used in a similar manner for all the examples of the present application. Similarly, mixtures of individual dyes can be used. The prints thus obtained have good fastnesses.

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APPLICATION EXAMPLE F

An ink consisting of 2.5 parts of the dyestuff of Example 94 in 97.5 parts of a mixture of water and 2-pyrrolidone where the ratio of water to N-methylpyrrolidone is 85:15 is introduced into an HP 880C DeskJet Printer and printed onto an A4 Epson Premium Glossy Photo Paper; S041287 (HP and DeskJet are registered trademarks of Hewlett-Packard, Palo Alto, California, USA; Epson is a registered trademark of Seiko Epson Kabushiki Kaisha.). The magenta prints thus obtained have very good ozon- and lightfastnesses.

15 This application example can be used in a similar manner for all the examples of the present application. Similarly, mixtures of individual dyes can be used. The prints thus obtained have good fastnesses

20 APPLICATION EXAMPLE G

An ink consisting of 2.5 parts of the mixture of Example 119 (dye mixture 1) in 97.5 parts of a mixture of water and 2-pyrrolidone where the ratio of water to N-methylpyrrolidone is 85:15 is introduced into an HP 880C DeskJet Printer and printed onto an A4 Epson Premium Glossy Photo Paper; S041287 (HP and DeskJet are registered trademarks of Hewlett-Packard, Palo Alto, California, USA; Epson is a registered trademark of Seiko Epson Kabushiki Kaisha.). The magenta prints thus obtained have very good ozon- and lightfastnesses.

This application example can be used in a similar manner for all the examples of the present application. Similarly, mixtures of individual dyes can be used. The prints thus obtained have good fastnesses

APPLICATION EXAMPLE H

An ink consisting of 2.5 parts of the mixture of Example 120 (dye mixture 2) in 97.5 parts of a mixture of water and 2-pyrrolidone where the ratio of water to N-methylpyrrolidone is 85:15 is introduced into an HP 880C DeskJet Printer and printed onto an A4 Epson Premium Glossy Photo Paper; S041287 (HP and DeskJet are registered trademarks of Hewlett-Packard, Palo Alto, California, USA; Epson is a registered trademark of Seiko Epson Kabushiki Kaisha.). The magenta prints thus obtained have very good ozon- and lightfastnesses.

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This application example can be used in a similar manner for all the examples of the present application. Similarly, mixtures of individual dyes can be used. The prints thus obtained have good fastnesses

CLAIMS

1. A dyestuff of formula (I)

$$Z_{1}$$
 Z_{2}
 Z_{3}
 Z_{3}
 Z_{3}
 Z_{3}
 Z_{3}
 Z_{3}
 Z_{4}
 Z_{5}
 Z_{3}
 Z_{5}
 Z_{3}
 Z_{4}
 Z_{5}
 Z_{5

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wherein

R₁ is H; C₁4alkyl; substituted C₁4alkyl; phenyl or substituted phenyl,

R₂ is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; -COOH; -COOCH₃; -CF₃; -SO₃H, -CN or SO₂NHR₆, where R₆ is H, C₁₋₄ Alkyl, phenyl or substituted phenyl

and

 X_1 is NR_3R_4 ; SR_5 ; OH;

 X_2 is NR_3R_4 ; SR_5 ; OH;

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R₃ is H, C₁₋₄alkyl; substituted C₁₋₄alkyl; phenyl or substituted phenyl, naphthyl or substituted naphthyl

R₄ is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; phenyl or substituted phenyl, naphthyl or substituted naphthyl

20 R₅ is C₁₋₄alkyl; substituted C₁₋₄alkyl; phenyl or substituted phenyl and

Z₁ is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; -OH; -COOH; -COOCH₃; -CF₃; -SO₃H; amino; alkylamino, -CN or SO₂NHR'₆, where R'₆ is H, C₁₋₄ alkyl, phenyl or substituted phenyl

25 Z_2 is H; C_{1-4} alkyl; substituted C_{1-4} alkyl; C_{1-4} alkoxy; OH; COOH; -SO₃H

Z₃ is H, C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; OH; COOH; -SO₃H as free acid or in salt form, as well as mixtures thereof.

2. A dyestuff according to claim 1 characterized in that

R₁ is H; C₁₋₄alkyl; substituted C₁₋₄alkyl,

R₂ is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; -COOH or -SO₃H and

5 X₁ is NR₃R₄; SR₅; OH;

 X_2 is NR_3R_4 ; SR_5 ; OH;

wherein '

R₃ is H, C_{1.4}alkyl; substituted C_{1.4}alkyl; phenyl or substituted phenyl, naphthyl or substituted naphthyl

is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; phenyl or substituted phenyl, naphthyl or substituted naphthyl

 R_5 is C_{1-4} alkyl; substituted C_{1-4} alkyl; phenyl or substituted phenyl and

is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; -OH; -COOH; -COOCH₃; -CF₃; -SO₃H; amino; alkylamino, -CN or SO₂NHR'₆, where R'₆ is H, C₁₋₄ alkyl, phenyl or substituted phenyl

Z₂ is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; OH; COOH; -SO₃H

 Z_3 is H, $C_{1\rightarrow}$ alkyl; substituted $C_{1\rightarrow}$ alkyl; $C_{1\rightarrow}$ alkoxy; OH; COOH; -SO₃H as free acid or in salt form, as well as mixtures thereof

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3. A dyestuff according to claim 2 characterized by the formula (Ia)

$$Z_2$$
 Z_3
 $N=N$
 $N=N$

wherein

R₁ is H; C₁₋₄alkyl; substituted C₁₋₄alkyl,

25 R₂ is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; -COOH or -SO₃H and

 X_1 is NR_3R_4 ; SR_5 ; OH;

 X_2 is NR_3R_4 ; SR_5 ; OH;

wherein

- R₃ is H, C₁₋₄alkyl; substituted C₁₋₄alkyl; phenyl or substituted phenyl, naphthyl or substituted naphthyl
- R₄ is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; phenyl or substituted phenyl, naphthyl or substituted naphthyl
- is C₁₋₄alkyl; substituted C₁₋₄alkyl; phenyl or substituted phenyl and
 - is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; -OH; -COOH; -COOCH₃; -CF₃; -SO₃H; amino; alkylamino, -CN or SO₂NHR'₆, where R'₆ is H, C₁₋₄ alkyl, phenyl or substituted phenyl
- is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; OH; COOH; -SO₃H

 Z₃ is H, C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; OH; COOH; -SO₃H

 as free acid or in salt form, as well as mixtures thereof
 - 4. A dyestuff according to claim 2 characterized by the formula (Ib)

$$Z_2$$
 $N=N$
 $N=N$

wherein

 R_1 is H; C_{1-4} alkyl; substituted C_{1-4} alkyl,

 R_2 is H; C_{1-4} alkyl; substituted C_{1-4} alkyl; C_{1-4} alkoxy; -COOH or -SO₃H and

20 X₁ is NR₃R₄; SR₅; OH;

 X_2 is NR_3R_4 ; SR_5 ; OH;

wherein

- R₃ is H, C₁₋₄alkyl; substituted C₁₋₄alkyl; phenyl or substituted phenyl, naphthyl or substituted naphthyl
- 25 R₄ is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; phenyl or substituted phenyl, naphthyl or substituted naphthyl

 R_5 is C_{1-4} alkyl; substituted C_{1-4} alkyl; phenyl or substituted phenyl and

is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; -OH; -COOH; -COOCH₃; CF₃; -SO₃H; amino; alkylamino, -CN or SO₂NHR'₆,

where R'6 is H, C1-4 alkyl, phenyl or substituted phenyl

Z₂ is H; C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; OH; COOH; -SO₃H

Z₃ is H, C₁₋₄alkyl; substituted C₁₋₄alkyl; C₁₋₄alkoxy; OH; COOH; -SO₃H as free acid or in salt form, as well as mixtures thereof

5. A process for the preparation of a compound according to the formula (I) characterized in that in a first step a compound of formula (II)

wherein all substituents have the meanings as defined above is reacted with a compound of formula (III)

leading to compounds according to formula (IV)

and in a second step the product of formula (IV) is reacted with one part of a compound of formula HX₁ wherein X₁ has the formula as described above which leads to compound of formula (V)

HO₃S
$$\frac{R_1}{N}$$
 $\frac{N}{N}$ $\frac{N}{$

and in a third step the compound of formula (V) is condensated with a compound of formula HX_2 wherein X_2 has the formula as described above leading to compound of formula (VI)

wherein substituents R₁ and R₂ have the same meanings as defined above and in a final step a compound of formula (VI) is coupled with the diazoniumsalt of a compound of formula (VII)

$$Z_1$$
 Z_2
 Z_3
 NH_2
(VII)

leading to the dyestuff of formula (I)

$$Z_1$$
 Z_2
 $N=N$
 $OH HN$
 R_2
 $N=N$
 $N=$

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wherein all substituents have the same meanings as defined above.

- 6. An Ink Jet Ink comprising at least one compound according to claim 1 or 2 or 3 or 4.
- 7. An Ink Jet Ink according to Claims 6 characterized in that the total content of salts is less than 0.5% by weight, based on the total weight of the dyes.

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- 8. Use of compounds according to claim 1 or 2 or 3 or 4 for printing recording material and/or in an inkjet printing process for printing recording materials and/or dyeing substrates comprising cellulose.
- 9. Use according to claim 8 characterized in that the recording material is paper or a papery substrate.
- 10. A recording material or a papery substrate or substrates comprising cellulose printed or dyed with a compound according to claim 1 or 2 or 3 or 4.

Disclosed are novel dyestuff of the formula (I)

$$Z_3$$
 Z_3
 $N=N$
 $N=N$

wherein all substituents are defined in the claims. These dyestuffs are useful for printing or dyeing subtrates, especially textile fibre materials, paper and papery substrates and plastic films and plastic transparencies.

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